

On behalf of our Department at UC-Berkeley, I would like to invite you to join us for the 10th Annual Eastman Foundation Distinguished Lectures in Catalysis, sponsored by the Eastman Foundation.

These lectures will be delivered virtually for this cycle in the form of a symposium featuring three leaders in academic catalysis research. Professors Regina Palkovits, Paul Dauenhauer, and Dave Flaherty will present their research and then share their thoughts about research in their fields in the form of a panel discussion.

This event will be held on 23 March 2021 at 9:30-11:30am PDT (12:30-2:30 pm EDT). Additional details are available at <https://chemistry.berkeley.edu/eastman-foundation-lectures>.

Registration is required via:

https://berkeley.zoom.us/webinar/register/WN_ug98bTSCSCaOuCVzkbuGMA.

I attach additional information and details below.

Regards

Enrique Iglesia
Theodore Vermeulen Chair in Chemical Engineering, University of California at Berkeley,
Laboratory Fellow, Pacific Northwest National Laboratory

**THE TENTH ANNUAL
EASTMAN FOUNDATION
DISTINGUISHED LECTURERS IN CATALYSIS**

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**March 23, 2021
9:00 am to 11:30 am Pacific Daylight Time**

For further information and registration instructions visit:
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or

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Regina Palkovits

Professor, RWTH Aachen University, Aachen/Germany
“Heterogeneous Catalysis and Process Design for closed CO₂ cycles”

Paul J. Dauenhauer

Professor, University of Minnesota
“The Catalytic Mechanics of Dynamic Surfaces for Energy Technology”

David W. Flaherty

Professor, University of Illinois at Urbana-Champaign
“Why Does the Catalyst Need to Be Wet? Activities, Active Sites and Non-Innocent Solvents”

Eastman Distinguished Visiting Lecturers

2010 – James Dumesic
2012 – Avelino Corma
2013 – Matthew Neurock
2014 – Bruce C. Gates
2015 – Mark Davis
2016 – Johannes Lercher
2017 – Jingguang Chen
2018 – Harold H. Kung
2020 – Susannah Scott

*Generous support for this visiting lecturer series is provided
by the Eastman Foundation*

“Heterogeneous Catalysis and Process Design for closed CO₂ cycles”

Regina Palkovits

Professor, RWTH Aachen University, Aachen/Germany

Renewable carbon feedstocks such as biomass and CO₂ present an important element of future circular economy. Especially biomass as highly functionalized feedstock provides manifold opportunities for the transformation into attractive platform chemicals. However, these resources require novel paradigms in process design. Fossil feedstocks are processed in stationary gas-phase processes at elevated temperature. On the contrary, biorefineries are based on processes in polar solvents at moderate conditions to selectively deoxygenate the polar, often thermally instable and high-boiling molecules. Considering “green electrons” provided by renewable energy technologies, also dynamic (electro)catalytic processes become attractive as key technology of a throughout circular economy. Herein, novel concepts in catalyst design will be discussed focusing on solid molecular catalysts for CO₂ activation, novel biomass transformations and the contribution of catalysts in life cycle assessment as well as the future role of a potentially electrified biorefinery.

“The Catalytic Mechanics of Dynamic Surfaces for Energy Technology”

Paul J. Dauenhauer

Professor, University of Minnesota

The emergence of competitive renewable energy from sunlight and wind heightens the importance of moving and storing energy from the rural places of origin to the locations where people live and work. Chemically capturing energy as compressed hydrogen or energy-dense liquids including ammonia or CO₂-derived methanol remains a leading method of energy storage based on density and fungibility, but the catalytic technology necessary for transforming electricity into chemicals in small distributed energy systems remains the key challenge. In this work, the general approach of dynamic catalyst operation is described as oscillatory binding energy of molecules adsorbed on surfaces as a method to dramatically accelerate the rate of catalytic reaction. Surface oscillations in sinusoidal and square waveforms of transient binding energy are imposed on catalyst surfaces with varying amplitude and frequency to identify the resonance conditions leading to order-of-magnitude enhancement in overall reaction rate. The results are presented in the context of catalyst-reaction behavior with regard to implementation in industrial reactor technologies necessary for moving and storing renewable energy.

“Why Does the Catalyst Need to Be Wet? Activities, Active Sites and Non-Innocent Solvents”

David W. Flaherty

Professor, University of Illinois at Urbana-Champaign

Solvent molecules surround and interact with catalytic active sites in ways that change reaction rates and selectivities by orders of magnitude. This seminar describes complex and previously unrecognized phenomena at solid-liquid interfaces during the direct synthesis of H₂O₂ (H₂ + O₂ → H₂O₂), an environmentally benign oxidant. H₂O₂ forms on Pd-based nanoparticles with structures that evolve in response to the reactant and solvent compositions. Alcoholic solvent molecules activate to form surface redox mediators that co-catalyze proton-electron transfer steps and influence the active phase of the catalyst nanoparticle. Water, in contrast, co-catalyzes and couples the hydrogen oxidation and oxygen reduction reactions on individual nanoparticles, analogous to electrochemical fuel cells. Comparisons of independent electrochemical and thermochemical measurements across alloy catalysts demonstrate fundamental connections between electro- and thermocatalytic processes for this chemistry. These concepts explain the dependence of rates and selectivities not only on operating conditions (temperature, reactant pressure, potential) but also on catalyst structure (e.g., Pd-Pd coordination). While we demonstrate these principles for an ostensibly simple reaction, such ideas may appear also for liquid-phase catalysis of other feedstocks.



Regina Palkovits is Full Professor for Heterogeneous Catalysis & Chemical Technology at RWTH Aachen University. She graduated in Chemical Engineering (Dipl.-Ing.) from Technical University Dortmund, Germany, in 2003. During studies, she spent a semester at Lehigh University, PA/USA. She carried out her PhD under supervision of Prof. Ferdi Schüth at the Max-Planck-Institut für Kohlenforschung (Mülheim an der Ruhr/Germany) until 2006. Afterwards, she joined the group of Prof. Bert Weckhuysen at Utrecht University, Netherlands, as postdoctoral fellow funded by a Hendrik Casimir-Karl Ziegler Fellowship. In 2008, she returned as a group leader to the Max-Planck-Institut für Kohlenforschung and since 2010 she is Professor at RWTH Aachen University. Her research focusses on the design of catalysts and processes for the efficient valorisation of renewable carbon resources. She published 170 articles in international peer-review journals, 9 books or book chapters and filed more than 25 patent applications. Professor Palkovits currently heads the Sustainable Chemistry Division of GDCh (German Chemical Society) and is vice-chair of the scientific advisory committee of Leibniz Institute for Catalysis (LiKat). She is Associate Editor for the RSC journal Catalysis Science & Technology, member of the national research building committee and core member of the Cluster of Excellence Fuels Science Center. She received numerous awards, including the 2019 EFCATS Young Researcher Award, the 2019 ExxonMobil Chemical European Science & Engineering Award, the 2016 DECHEMA Award, and the Innovation Prize of the State of North Rhine-Westphalia in the Young Researchers category. Since 2019, she is a Max Planck Fellow at the Max Planck Institute for Chemical Energy Conversion and as of 2020, a member of the North Rhine-Westphalian Academy of Sciences, Humanities and the Arts.



Paul Dauenhauer received a bachelor of science in chemical engineering and chemistry from the University of Wisconsin in 2004 and a Ph.D. in chemical engineering from the University of Minnesota in 2008 supervised by Professor Lanny D. Schmidt. From 2008 to 2009, Paul worked as a senior research engineer for the Dow Chemical Company within Core R&D Reaction Engineering in Midland, MI, and the Hydrocarbons & Energy group in Freeport, TX. In 2009, he joined the University of Massachusetts Department of Chemical Engineering as an assistant professor. Since joining the University of Minnesota Department of Chemical Engineering & Materials Science in 2014, he has served as the Lanny Schmidt Honorary Professor, MacArthur Fellow, and the co-Director of the Catalysis Center for Energy Innovation. His published patent applications serve as the scientific foundation of three startup companies: Sironix Renewables, Activated Research Company, and enVerde, LLC.



David Flaherty earned a bachelor of science in chemical engineering from the University of California at Berkeley and a Ph.D. in chemical engineering from the University of Texas at Austin with Professor Buddie Mullins. He was a postdoctoral scholar at the University of California, Berkeley with Prof. Enrique Iglesia. Since 2013, joining as the Dow Chemical Company Faculty Scholar and Associate Professor of Chemical and Biomolecular Engineering at the University of Illinois, Flaherty leads a group that develops understanding and design principles for the use of solid catalysts to resolve challenges for the sustainable production of chemicals and energy carriers. His research focuses on generating new insight to the chemical phenomena that emerge when reactions occur at complex surfaces and at liquid-solid interfaces. Prof. Flaherty is the recipient of the DOE Early Career Award (2019), NSF CAREER Award (2016), the AVS, Prairie Chapter, Early Career Research Award (2018), the Dean's Award for Research Excellence from the College of Engineering (2018), and the Doctoral New Investigator Award of the ACS-PRF (2014), in addition to recognition for his contributions as a teacher and advisor.

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March 23, 2021 (9:00-11:30 am)

Opening Remarks

9:00 am

Alexis T. Bell, Enrique Iglesia, Alexander Katz

Introduction of Regina Palkovits by Alexis T. Bell

9:02 am

“Heterogeneous Catalysis and Process Design for Closed CO₂ Cycles”

Regina Palkovits

Presentation: 9:05 am

Q&A: 9:35 am

Introduction of Paul Dauenhauer by Alexander Katz

9:42 am

“The Catalytic Mechanics of Dynamic Surfaces for Energy Technology”

Paul J. Dauenhauer

Presentation: 9:45 am

Q&A: 10:15 am

Introduction of David Flaherty by Enrique Iglesia

10:22 am

**“Why Does the Catalyst Need to Be Wet?
Activities, Active Sites and Non-Innocent Solvents”**

David W. Flaherty

Presentation: 10:25 am

Q&A: 10:55 am

Introduction to Panel Discussion by Alexis T. Bell

11:02 am

“Future Research Directions in Catalysis”

One-Slide Presentation by Panelists

11:04 am

Open Panel Discussion

11:10 am

Closing Remarks by Alexis T. Bell

11:29 am

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